Material Safety Data Sheet
Cyanogen bromide

MSDS\# 84299
Section 1 - Chemical Product and Company Identification
MSDS Name: Cyanogen bromide
Catalog AC110780000, AC110780050, AC110785000, AC405950000, AC405950250, AC405951000
Numbers: AC405951000, AC405955000, 11078-0250, 11078-1000, O6103-100, O6103-25
Synonyms: Bromocyanide.

| Company Identification: | Fisher Scientific <br> One Reagent Lane <br> Fair Lawn, NJ 07410 |
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| For information in the US, call: | $201-796-7100$ |
| Emergency Number US: | $201-796-7100$ |
| CHEMTREC Phone Number, US: | $800-424-9300$ |

Section 2 - Composition, Information on Ingredients

CAS\#:
Chemical Name:
\%:
EINECS\#:

Hazard Symbols:


Risk Phrases:

506-68-3
Cyanogen bromide
97+
208-051-2
$\mathrm{T}+\mathrm{N}$


26/27/28 3450

Section 3 - Hazards Identification

## EMERGENCY OVERVIEW

Danger! Lachrymator (substance which increases the flow of tears). May cause cardiac disturbances. May cause pulmonary edema. May cause cyanosis (bluish discoloration of skin due to deficient oxygenation of the blood). Causes burns by all exposure routes. May be fatal if inhaled, absorbed through the skin or swallowed. Material is a solid at room temperature that melts upon moderate heating into a combustible liquid with a flash point below $200^{\circ} \mathrm{F}\left(93.3^{\circ} \mathrm{C}\right)$. Very toxic to aquatic organisms. Target Organs: Blood, central nervous system, respiratory system, gastrointestinal system, eyes, skin. Potential Health Effects
Eye: Causes eye burns. Lachrymator (substance which increases the flow of tears).
Skin: May be fatal if absorbed through the skin. Causes skin burns. May be metabolized to cyanide which in turn acts by inhibiting cytochrome oxidase impairing cellular respiration. Substance is readily absorbed through the skin. May be fatal if swallowed. Causes gastrointestinal tract burns. May cause severe gastrointestinal tract irritation Ingestion: with nausea, vomiting and possible burns. Metabolism may release cyanide, which may result in headache, dizziness, weakness, collapse, unconsciousness and possible death.
May be fatal if inhaled. May cause cyanosis (bluish discoloration of skin due to deficient oxygenation of the blood). Causes chemical burns to the respiratory tract. May cause dyspnea (difficult or labored breathing). May cause nausea, dizziness, and headache. May produce cardiovascular effects. May be metabolized to cyanide Inhalation: which in turns act by inhibiting cytochrome oxidase impairing cellular respiration. Inhalation may result in symptoms similar to cyanide poisoning which include tachypnea, hyperpnea (abnormally rapid or deep breathing), and dyspnea (labored breathing) followed rapidly by respiratory depression. Pulmonary edema may occur. May cause chest pain and lung irritation.

Chronic exposure to cyanide solutions may lead to the development of a "cyanide" rash, characterized by itching, and by macular, papular, and vesicular eruptions, and may be accompanied by secondary infections. Exposure to small amounts of cyanide compounds over long periods of time is reported to cause loss of appetite, headache, weakness, nausea, dizziness, and symptoms of irritation of the upper respiratory tract and eyes.

## Section 4 - First Aid Measures

Eyes:

Skin:

Ingestion:

Inhalation:

Notes to
Physician:

Immediately flush eyes with plenty of water for at least 15 minutes, occasionally lifting the upper and lower eyelids. Get medical aid immediately.
POISON material. In case of contact, get medical aid immediately. Immediately flush eyes or skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Wash clothing before reuse. Destroy contaminated shoes.
POISON material. If swallowed, get medical aid immediately. Only induce vomiting if directed to do so by medical personnel. Never give anything by mouth to an unconscious person.
Remove from exposure and move to fresh air immediately. If breathing is difficult, give oxygen. SPEED IS ESSENTIAL, OBTAIN MEDICAL AID IMMEDIATELY. Do not use mouth-to-mouth resuscitation if victim ingested or inhaled the substance; induce artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device.

## Section 5 - Fire Fighting Measures

As in any fire, wear a self-contained breathing apparatus in pressure-demand, MSHA/NIOSH (approved

General Information: or equivalent), and full protective gear. Combustion generates toxic fumes. Will burn if involved in a fire. Containers may explode in the heat of a fire. This chemical poses an explosion hazard. Material is a solid at room temperature that melts upon moderate heating into a combustible liquid with a flash point below $200^{\circ}$ $\mathrm{F}\left(93.3^{\circ} \mathrm{C}\right)$.
Extinguishing Media:

Use carbon dioxide or dry chemical. DO NOT USE WATER OR FOAM.
Autoignition
Temperature:
Not applicable.
Flash Point: > $65 \operatorname{deg} \mathrm{C}(>149.00 \operatorname{deg} \mathrm{~F})$
Explosion
Limits: Lower:
Not available
Explosion Limits: Upper: available
NFPA Rating: health: 4; flammability: 2; instability: 3;
Section 6 - Accidental Release Measures
General Information:

Use proper personal protective equipment as indicated in Section 8.
Vacuum or sweep up material and place into a suitable disposal container. Wear a self contained breathing Spills/Leaks: apparatus and appropriate personal protection. (See Exposure Controls, Personal Protection section). Avoid generating dusty conditions. Remove all sources of ignition. Use a spark-proof tool. Evacuate unnecessary personnel. Do not let this chemical enter the environment.

## Section 7 - Handling and Storage

Minimize dust generation and accumulation. Use spark-proof tools and explosion proof equipment. Do not get in Handling: eyes, on skin, or on clothing. Keep away from heat, sparks and flame. Do not ingest or inhale. Container should be opened by a technically qualified person. Use only in a chemical fume hood.

Storage:
Keep away from sources of ignition. Store in a tightly closed container. Store in a dry area. Keep refrigerated. (Store below $4^{\circ} \mathrm{C} / 39^{\circ} \mathrm{F}$.) Store protected from moisture. Store protected from light.

Section 8 - Exposure Controls, Personal Protection


Engineering Controls:
Use explosion-proof ventilation equipment. Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower. Use only under a chemical fume hood.

## Exposure Limits

Personal Protective Equipment
Eyes: Wear appropriate protective eyeglasses or chemical safety goggles as described by OSHA's eye and face protection regulations in 29 CFR 1910.133 or European Standard EN166.
Skin: Wear appropriate protective gloves to prevent skin exposure.
Clothing: Wear appropriate protective clothing to prevent skin exposure.
Respirators: A respiratory protection program that meets OSHA's 29 CFR 1910.134 and ANSI Z88.2 requirements or European Standard EN 149 must be followed whenever workplace conditions warrant respirator use.

## Section 9 - Physical and Chemical Properties

Physical State: Crystals
Color: white
Odor: pungent odor
pH : Not available
Vapor Pressure: 116 mbar @ 20 deg C
Vapor Density: Not available
Evaporation Rate: Not available
Viscosity: Not available
Boiling Point: 61-62deg C @ 760 mmHg
Freezing/Melting Point: 49-54 deg C
Decomposition Temperature: Not available
Solubility in water: Decomposes.
Specific Gravity/Density: 2.015
Molecular Formula: CBrN
Molecular Weight: 105.93
Section 10 - Stability and Reactivity

Chemical Stability:

Conditions to Avoid:
Incompatibilities with Other Materials
Hazardous Decomposition
Products
Hazardous Polymerization

Unstable. May decompose on exposure to moist air or water. May undergo autopolymerization. Moisture sensitive. Light sensitive.
Incompatible materials, light, ignition sources, dust generation, metals, exposure to moist air or water.

Metals, oxidizing agents, acids, alcohols, amines, ammonia.
Hydrogen cyanide, nitrogen oxides, hydrogen bromide.
May occur.

Section 11 - Toxicological Information
RTECS\#: CAS\# 506-68-3: GT2100000
LD50/LC50: RTECS: Not available.
Carcinogenicity: Cyanogen bromide - Not listed as a carcinogen by ACGIH, IARC, NTP, or CA Prop 65.
Other: $\quad$ The hazards associated with cyanide may be seen in this product.
Section 12 - Ecological Information
Other: Do not empty into drains.

## Section 13 - Disposal Considerations

Dispose of in a manner consistent with federal, state, and local regulations.
Section 14 - Transport Information
US DOT
Shipping Name: CYANOGEN BROMIDE
Hazard Class: 6.1

UN Number: UN1889
Packing Group: I
Canada TDG
Shipping Name: CYANOGEN BROMIDE
Hazard Class: 6.108
UN Number: UN1889
Packing Group: I

USA RQ: CAS\# 506-68-3: 1000 lb final RQ; 454 kg final RQ
Section 15 - Regulatory Information

## European/International Regulations

## European Labeling in Accordance with EC Directives

Hazard Symbols: T+N
Risk Phrases:
R 26/27/28 Very toxic by inhalation, in contact with skin and if swallowed.
R 34 Causes burns.
R 50 Very toxic to aquatic organisms.
Safety Phrases:
S 26 In case of contact with eyes, rinse immediately with plenty of water and seek medical advice.
S 28A After contact with skin, wash immediately with plenty of water.
S 36/37/39 Wear suitable protective clothing, gloves and eye/face protection.
S 45 In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).
S 61 Avoid release to the environment. Refer to special instructions/safety data sheets.
WGK (Water Danger/Protection)
CAS\# 506-68-3: 3
Canada
CAS\# 506-68-3 is listed on Canada's DSL List
Canadian WHMIS Classifications: B3, D1A, E, F
This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all of the information required by those regulations.
CAS\# 506-68-3 is listed on Canada's Ingredient Disclosure List

## US Federal

TSCA
CAS\# 506-68-3 is listed on the TSCA
Inventory.
Section 16 - Other Information
MSDS Creation Date: 6/02/1998
Revision \#7 Date 7/20/2009

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